

Review of ICARE-CNRS activities on facilitating CO₂ capture from the combustion of fossil fuels

Centre National de la Recherche Scientifique - CNRS Institut de Combustion, Aérothermique, Réactivité et Environnement - ICARE

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Institut de Combustion, Aérothermique, Réactivité et Environnement

Outline

- Few words about ICARE-CNRS @ Orléans
- Basic chemical kinetic studies on the oxydation of methane, ethylene and propene in the presence of CO2 and H2O (ANR project TACOMA)
- Combustion of methane in oxygene enriched air coupled with membrane capture of CO2. Basic and pilot system studies (CNRS Energy Program project COCASE)
- Retrofitting a pulverised coal power plant into a oxyfuel system. (FP7 TREN demonstration project SOMALOX)
- Interdisciplinary and comparative analyses of the acceptability of new energy technologies: the case studies of CCS in France and Canada (CNRS Energy Program project ALICANTE)
- 10th ICCEU

ICARE in Orléans

- Institut de
- Combustion
- Aérothermique
- Réactivité et
- Environnement
- Institute for Combustion, Aerothermal sciences, Reactivity, Environnement





Orléans ICARE is in Orléans, 125 km from Paris

ICARE - CNRS Institut de Combustion, Aérothermique Réactivité et Environnement 1c, avenue de la Recherche Scientifique 45071 Orléans - Cedex 2 - France

Total staff: 120 26 Researchers and faculty 21 Engineers and technicians 39 PhD students and post-docs 34 Various trainees

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Two main research domains:

Energy & Environnement Space & Propulsion

Four main research thematics:

Chemical kinetics of combustion and reactive systems Dynamics of combustion and reactive systems Atmospheric chemistry Supersonic, hypersonic, rarefied, ionized flows

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Research domains of ICARE

Energy & Environment Propulsion & Space

- Combustion
- Chemical kinetics
- Plasmas physics
- Fluid mechanics, turbulence
 - Two phase flows
 - Supersonic, hypersonic flows
 - Ionized, rarefied flows







- Aerospace propulsion
- Electric propulsion
- Liquid and solid propulsion
- Atmospheric reentry
- Atmospheric chemistry
- Energy production
- Alternative fuels, biofuels, hydrogen
- Pollutant emissions reductions
- Industrial risk prevention







Main international cooperations: UE, Russie, USA, Canada, Chine, Japon, Ukraine, Turquie, Argentine, etc

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Project TACOMA (GDF SUEZ; TOTAL; IFP; ICARE-CNRS)

- A new combustion mode for glass, petrochemistry, steel industries to concentrate CO2 in the flue gases
- Combustion of heavy oil and natural gas diluted with recirculation of burnt gases
- Ethylene and propene are the main products of the cracking of heavy oil
- Oxydation of C2H4 and C3H6 in O2/N2/CO2/H2O at atmospheric pressure, over a wide range of initial concentrations, temperatures and equivalence ratios. A detailed chemical kinetic reaction mechanism was used

to simulate the experiments and analyze the results.

Project TACOMA-ICARE CNRS Conclusions

- The oxidation of ethylene is inhibited by water and the effect decreases with the increasing equivalence ratio. Carbon dioxide accelerates slightly the consumption of C2H4 under fuel-rich conditions. The oxidation of ethylene in presence of water vapor yields reduced formation of carbon monoxide and increased acetylene production. Carbon dioxide participates to the production of CO when initially present in ethylene-O2-N2 reacting mixtures.
- The oxidation of propene is less affected by the presence of CO2 or H2O in the reacting mixture under the present JSR conditions.

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Oxygen enriched air combustion of methane in relation with membrane capture of CO2 Project COCASE (CNRS Energy Program)

- Membrane capture of CO2 is a promising technology but needs a minimum of 20% CO2 concentration in the flue gases.
- Combustion of CH4 in oxygen enriched air with CO2 recirculation is investigated to optimise the O2 concentration, the CO2 recirculation rate, the CO2 concentration rate and the membrane properties

Project COCASE: Global principle



Impact énergétique de la récupération du CO₂

Cycle turbine réaliste + recyclage + recompression - capture $CO_2(idéal)$

: $P_3 = HP = 30bar$, $P_1 = P_2 = P_{atmo}$, $T_4 = 1200 \,^{\circ}C$

 $\eta_{is\text{-compresseur}} = 0.9, \quad \eta_{is\text{-turbine}} = 0.9, \ \Delta P_{Ch\ Comb} = 5\%$



Impact énergétique de la récupération du CO2

Impact du taux de recirculation Z et de la pression P_{in} en entrée du séparateur sur le rendement



Impact énergétique de la récupération du CO2

Impact du taux de recirculation Z et de la pression P_{in} en entrée du séparateur sur le cout de capture du CO₂



Impact énergétique de la récupération du CO2

Taux de récupération de la membrane

en fonction de la sélectivité de la membrane et de la pression Pin (entrée membrane)



Résultats à phi=0,9 (condition n°1 : enrichissement seul) Température et vitesse de flamme en f° du %O₂ dans l'air (1, 4, 8 bars), T_o=600K



Résultats à phi=0,9 :

Température et vitesse de flamme en f° du $%O_2$ dans l'air, T_o=600K, avec ou sans dilution (4 bars)



Résultats à phi=0,9 :

 CO_2 et NOx (Thomsen) en f° du % O_2 dans l'air, T_0 =600K, avec ou sans dilution (4 bars)



Résultats à phi=0,9 :

Refroidissement : environ 60% de CO_2 froid (300K) nécessaire quelle que soit la pression et les conditions



High Pressure Chamber

- H = 1.2 m
- Dint = 0.3 m
- Water cooling system
- Windows heating system
- Laser light absorbing paint

Axial displacement



Laminar Burner



Turbulent Burner

Experimental conditions

- $U \approx 2.1 \text{ m/s}$
- $\phi = 0.6 \text{ à } 0.7$
- P = 0.1 à 0.9 MPa
- Pilot flame flow < 7% Main flow
- T = 300 K
- u' et L_U cste



Flow conditions

- Mean flow velocity :
 - $U_0 = 0.7$ to 1 m/s (Laminar)
 - $U \approx 2.1 \text{ m/s}$ (Turbulent)
- Equivalence ratio :
 - $\phi = 0.9$ to 1.05 (Laminar)
 - $\phi = 0.6$ (Turbulent)
- CO₂ addition :

$$(1-\beta)[CH_4 + (2+x)(O_2 + 3.78N_2)] + \beta CO_2$$

$$\rightarrow CO_2 + 2(1-\beta)H_2O + 7.56(1-\beta)N_2 + x(1-\beta)(O_2 + 3.78N_2)$$

where $\beta = \frac{n(CO_2)}{n(CH_4) + n(CO_2)}$ and x the air excess

- Pressure :
 - P < 0.2 MPa (Laminar)
 - P < 0.9 MPa (Turbulent)
- Temperature of fresh gases : 300 K

Laser diagnostics



Methodology

- ➡ Camera CCD : 40 Hz Resolution : 34.8 µm/pixel
- ▶ 500 images for each experimental condition

Pressure effect

 $(U_0=1 \text{ m/s et } \phi=1)$



P = 0,1 MPa P = 0,125 MPa P = 0,15 MPa P = 0,175 MPa P = 0,2 MPa

<u>CO₂ addition effect</u> (P=0.1MPa - ϕ =0.95 -U₀=1.2m/s)



Laminar flame velocities



Flame surface density



BML Model





- Mean fuel consumption rate decrease with the dilution rate
- Power at 0.1 MPa = 2.2 kW (theoretical = 2 kW)
- Power at 0.9 MPa = 12 kW
 (theoretical = 14 kW)

Gas turbine combustion chamber facility of CORIA-CNRS, Rouen



 $CH_4 + (1+e) 2 (O_2 + 3.78 N_2 + b CO_2)$

CNRS – UNIVERSITE et INSA de Rouen

Evolution of turbulent flame structure with CO2 dilution rate, at contant equivalence ratio

b = 0 b = maxExtinction Lifted flame **Attached flame** CNRS – UNIVERSITE et INSA de Rouen

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<u>Topic ENERGY.2008.6.1.3: Efficiency Improvement of Oxygen-based</u> <u>combustion</u>

Content/scope: Further research and demonstration work is needed on oxygen based combustion technologies in respect to the CO2 capture process to make this technology available for large scale power plants. Work in this area should include – but not be limited to - issues like advanced burner designs as well as slagging, fouling and corrosivity of flue gases, the identification of the optimal CO2 capture rate, combination of CO2 capture with other gas cleaning processes and processes for separation, compression and conditioning of CO2. It is envisaged that a project under this topic will test, demonstrate and further develop existing technologies in a medium sized test environment. Scalability of the results to large scale power plants has to be in the focus of the activities. *Funding scheme:* Collaborative project with a dominant demonstration component.

Expected impact: Oxygen-based combustion technologies can play an important role for CCS. Projects under this topic shall further develop these technologies and test them in small scale demonstration plants and thereby pave the way for their use in industrial scale power plants. *Open in call:* FP7-ENERGY-2008-2

Why oxy-fuel lignite combustion ?

- To improve the efficiency of lignite combustion; retrofitting possible; rapid benefits for electricity production for countries such as Turkey and Greece
- To ease CO2 capture
- less NOx as recycled NO (fuel NO) is burned out and no thermal NOx (no N2)
- EOR
- Enhanced CBM

Issues (1)

Retrofit is possible and demonstrated in pilot scale systems. O2 is mixed with a fraction (important about 65%) of the flue gases (CO2 and H2O mainly or if after condensation mainly only CO2). This helps to reduce the flame temperature and obtain flame conditions equivalent to air burning. As the CO2 and H2O thermal capacities are higher than that of N2, more O2 concentration is needed to obtain full conversion. It is possible to adjust the O2 and flue gas concentration to match the heat transfer parameters to that of air burning

Issues (2)

- As the flow physical parameters are different between air and recycled flue gas and oxygen mixture, if we want to match the heat transfer rate, the flow rates will be different. Therefore a new burner design is necessary
- SO2 SO3 emissions may increase, corrosion should be handled
- Heat transfer parameters may also need adjustment; no air in leakage should be allowed, so the boiler should be very well sealed

Issues (3)

- ASU (air separation unit) is the most energy consuming part, 95% O2 is generally used to reduce the cost penalty
- Also for CO2 recycling and capture compressors are necessary
- Therefore, oxy-fuel coal burning with CCS gives clearly a lower energy efficiency compared to air burning
- But CCS with air burning and amine based systems is even more costly; then we have to compare comparable issues

Issues (4)

- The efficiency issue should be considered globally. With oxy-fuel combustion we shall burn much more efficiently the lignite. Combustion studies should be done
- Captured CO2 should be used intelligently: for EOR, for ECBM, for inclusion into cement and concrete production, for ex-situ carbonation etc. We have to increase the value of the whole chain by using CO2 as a commodity

Issues (5)

- We need to chose a demonstration site for oxy-fuel lignite combustion: SOMA A: EUAS, TKI
- Compatible power level with the FP7 requirements: 22 MW with several burners of the MW level
- Pulverized lignite with oxy-fuel retrofitting, dissemination and scalability potential
- Enhanced CBM potential in the area...

SOMALOX

- A proposal for FP7 ENERGY TREN 2008 to demonstrate retrofitting the SOMA A pulverized lignite power plant (22 MW) to oxy-fuel combustion
- Partnership: AEE, RWE NPower, CKD Export, ENEA, TKI, EUAS, HABAS, CNRS, NTUA, JRC-IE, Cottbus, Nottingham...

Where is SOMA ? 150 km north-east of Izmir

SOMALOX Work package structure

- WP I: Simulation of the optimisation of the SOMA A power plant for oxy-fuel lignite combustion retrofitting Simulation of the global process
 Optimisation of the lignite preparation for oxy-fuel combustion
- WP II Optimisation of oxygen production, gas cleaning, CO2 capture
- WP III Optimisation of the oxy-fuel lignite burner
- WP IV Retrofitting and Demonstration at the plant level
- WP V Economic, Environmental, Dissemination, Scalability analysis

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THE NINTH ASIA-PACIFIC INTERNATIONAL SYMPOSIUM ON COMBUSTION AND ENERGY UTILIZATION (9th APISCEU)

> November 02-06, 2008 Beijing, China

Organized by Beijing Society of Thermophysics and Energy Engineering, China Institute of Engineering Thermophysics, Chinese Academy of Sciences Beijing Shenwu Thermal Energy Technology Co, LTD, China Yanshan Petro-Chemical Industry Corporation, China The Combustion Institute, USA Sponsored by Beijing Association for Science and Technology, China

Internet: http://www.APISCEU9.org.cn

10th ICCEU

- 10th International Conference on Combustion and Energy Utilisation
- Will be organized at the University of Mugla, Turkey, on 4-8 May 2010
- Main topic: Clean fossil fuel technologies : (technical and socio-economic aspects)
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